

Determining Dilution Series

This technote provides information about the concentration range spanned and preparation calculations for serial dilutions. *Figure 1* illustrates the range of concentrations spanned by a series of twelve dilutions with commonly utilized dilution factors. The following provides calculation for preparation of a serial dilution factor. **Bold** indicates standard calculations regardless of the dilution factor used. Refer to *Figure 2* to clarify the experimental set up.

Sample Experiment Example:

- 3 mL Sample Volume
- 2.5 fold serial dilution

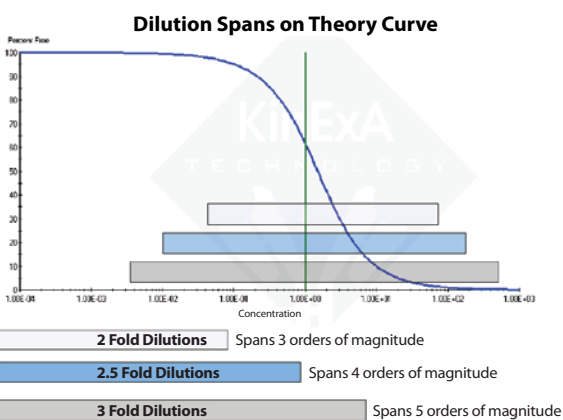


Figure 1.

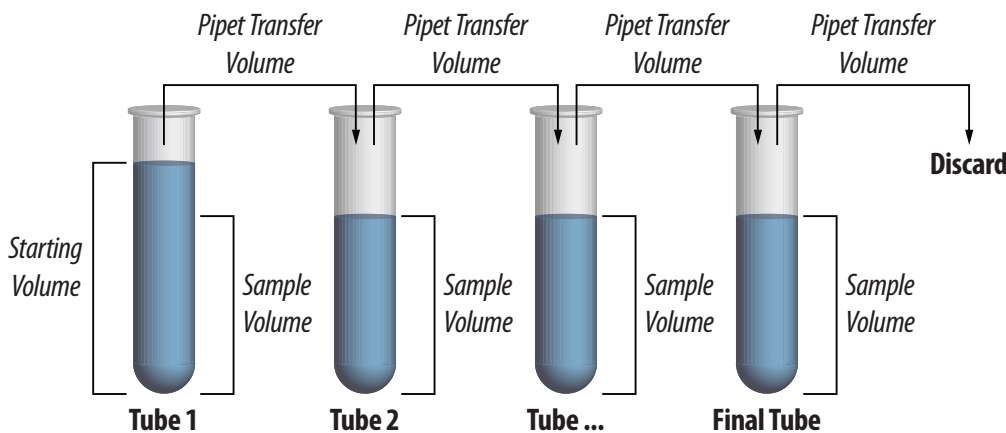


Figure 2.

To Prepare Serial Dilutions:

- Calculate the total *Sample Volume* necessary to run each sample.
- Calculate the *Starting Volume* for tube 1:

$$\text{Starting Volume} = \frac{\text{Sample volume} \times \text{Dilution Factor}}{(\text{Dilution Factor} - 1)}$$

$$\text{Starting Volume} = (3 \text{ mL} \times 2.5) / (2.5 - 1)$$

$$\text{Starting Volume} = 5 \text{ mL}$$

- Calculate the *Pipet Transfer Volume*:

$$\text{Pipet Transfer Volume} = \text{Starting Volume} - \text{Sample Volume}$$

$$\text{Pipet Transfer Volume} = 5 \text{ mL} - 3 \text{ mL}$$

$$\text{Pipet Transfer Volume} = 2 \text{ mL}$$